CORRECTION

High Bulk Electron Mobility Diketopyrrolopyrrole Copolymers with Perfluorothiophene

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Table 2 of the article was erroneously published with an extra column. The correct table can be found below. The editors apologize for any confusion.

Table 2. Solid state electrochemical and structural properties of the PDPPs.

Polymer	Cyclic Voltammetry				XRD			
	IP[a] [eV]	EA[b] [eV]	E_{CV}[c] [eV]	E_{opt}[d] [eV]	d_{lam}[e] [Å]	d_{amorph}[f] [Å]	$d_{\pi-\pi}$[g] [Å]	ζ[h] [nm]
PDPP[T]$_2$-T	–5.75	–3.69	2.06	1.32	19.16	4.52	3.86	6.9
PDPP[T]2-TF2	–5.91	–3.70	2.21	1.33	19.83	4.40	3.65	8.8
PDPP[Py]$_2$-T	–6.08	–3.78	2.30	1.72	19.05	4.56	3.91	3.9
PDPP[Py]$_2$-TF$_2$	–6.11	–3.81	2.30	1.73	18.63	4.48	3.90	4.1
PDPP[Ph]$_2$-T	–5.97	–3.53	2.44	1.98	14.94	4.04	–	2.7
PDPP[Ph]$_2$-TF$_2$	–6.04	–3.57	2.47	2.03	14.62	3.90	–	2.1

[a]CV based ionization potential; [b]CV based electron affinity; [c]Electrochemical gap determined by CV in thin film; [d]Optical gap determined from absorption onset in film; [e]Lamellar spacing; [f]Amorphous peak; [g]π–π spacing; [h]Crystal correlation length $\zeta = 2\pi/\text{FWHM}$ of the lamellar spacing.

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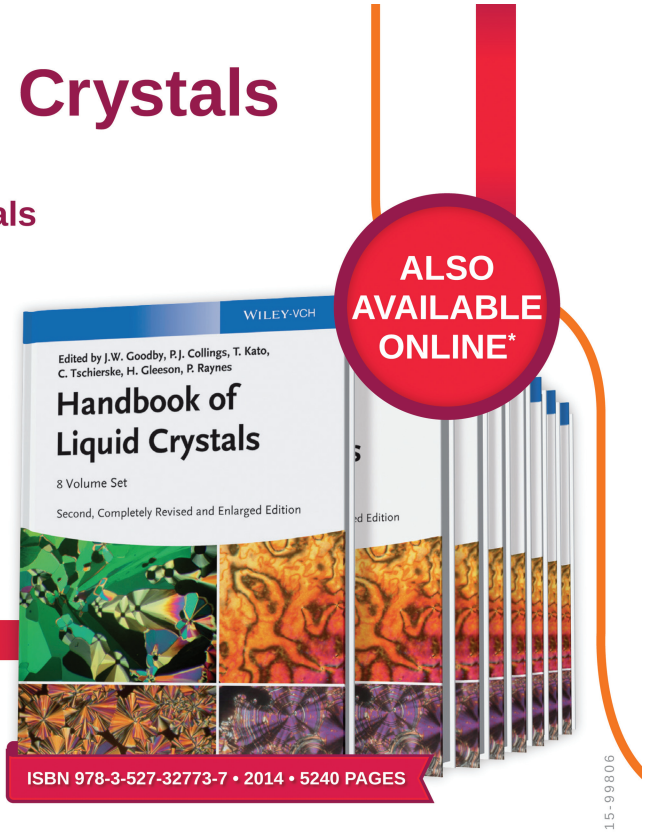

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